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Spatial and Temporal Water Quality Evaluation of Heavy Metals of Habbaniyah Lake, Iraq

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## ABSTRACT

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### Keywords:

water quality, pollution assessment, heavy metal, heavy metal pollution index (HMPI), heavy metal evaluation index (HMEI), contamination degree (CD)

Heavy metal concentrations in a water sample from Habbaniyah Lake (HL), Iraq (Cr, Cd, Ni, Fe, Co, Zn, Pb, and Cu). Thirty-three sites were chosen in the research area from October 2020 to April 2021 to evaluate emissions of heavy metals during two seasons (dry and wet). Spectrophotometer for atomic absorption was used to test heavy metal concentration (USA Phoenix-986). The findings of the study indicate that water levels were concentrated Cr>Ni >Fe>Co>Pb>Zn=Cd=Cu during the dry season and the wet season Cr>Fe>Pb>Co>Ni=Zn=Cd=Cu. HMPI and HMEI were both used to assess the level of water supply toxicity of heavy metals in the area of research. In Habbaniyah Lake water, several concentrations of heavy metals exceed the criteria for drinking and water life such as Co and Pb were exceed the allowed limits of WHO, CCME in the dry and wet seasons. The study area's human, agricultural and industrial activities and human population resulted in a rise in heavy element concentrations including Cr that surpassed the WHO, CCME, FAO, EPA, and IQS. The mean values of (Cd, Zn, and Cu) during dry and wet seasons were non-consistent in all stations. The average value of Ni was above water life limit during the dry season. The average values of CD, HMPI, and HMEI during the wet season were 20.8984, 57.8248 and 24.8977 and in the dry season were 17.3745, 61.8769, and 22.3747 respectively. Results indicate that HL is highly contaminated with HMs according to national and international guidelines, (CD), (HMPI) and (HMEI) indices pointed that HL water quality was bad. Principal component analysis (PCA) and cluster analysis (CA) were applied to estimate the pollution sources, results show that, pollution are originated from multiple sources, anthropogenic sources are major pollution sources while lithogenic is minor pollution sources, anthropogenic origin, which is mostly due to the wastewater point sources on Warrar stream.

# 1. INTRODUCTION

The nature of the water is famous for its physical, chemical, and biological properties [1]. Due to their toxicity and alleged persistence within aquatic settings, interest in heavy metals (HMs) in environmental conditions has increased in past years [2, 3]. Heavy metals, which include both vital and nonessential elements, are particularly important in ecotoxicology because they are very persistent and can all be harmful to live species [4-6]. Trace metals are among the most important environmental toxins and the presence of natural or biota in water sources of anthropogenic [7]. Surface water can pose a major danger to water environments through chemical toxicity causing acute and chronic toxicity in water environments, ecosystem accumulation, habitat degradation, and biodiversity destruction, as well as a potential danger to human health [8]. Trace metals have taken place in marine habitats and have created significant questions over their effects on plants and animals. Heavy metals accumulation in water, the clean safe water, which another of the main elements for human wellbeing, induces water loss and thus reduces the quality of water and cannot use the water in some uses [9]. The toxicity of heavy metal for people and the ecosystem is a concern because it accumulates in the food chain and is nonbiodegradable. Metals like Fe, Zn, Cu, and Ni are essential for animals and plants to survive, but other metals like Pb, Cd, and Cr have physiological effects on animals and plants. To evaluate the state of pollution in the environment, heavy metal concentrations can also be analyzed and distributed in the environment. [5, 10, 11]. Heavy metals are most much the source of pollution of the water system because of numerous anthropological causes such as industrial wastewater, drainage of municipal surface water, agriculture processes, etc. [12]. Sources of heavy metals are anthropogenic (agriculture activities, industries, and mining) and natural (weathering of soils and rocks) [8, 11, 13-18]. These heavy metals can be accumulated strongly along the water, the food chain, and sediment which causes the deadly effects of aquatic organisms [19, 20]. Natural sources are mostly low in concentrations, however higher levels of anthropogenic sources cause environmental problems [3]. Agricultural, human and industrial activities are the main cause of toxic metals accumulates in rivers, sediments, plants, and animals [21-23]. Quality of water is decreasing due to indiscriminate use of pesticides and heavy-metal fertilizers in irrigation, mining, final disposal of toxic metal-bearing waste and metal from various factories such as steel, battery production, thermal energy, and etc. [24]. Several water quality assessing approaches, including the Pollution Index (CD), (HPI) and the Heavy Metal Assessment Index have been established (HEI). Al-Saadi and et al developed Regression Model for Estimating Irrigation Water Quality Index, the data demonstrate that 90 percent of groundwater wells have IWQI levels that fall into the Severe Restriction category, while the remaining 10% had IWQI values that fall into the High Restriction category. The constructed model's calibration and validation revealed that it has a decent estimation of the IWQI values [25].

The actual water discharge levels are measured in these indexes and integrate all parameters of water pollution into a single method [17]. This research aims at determining heavy metal concentrations (Cr, Cd, Ni, Fe, Co, Zn, Pb, and Cu) in water of Habbaniyah Lake, estimating the Heavy Metal Pollution Index (HMPI), Heavy metal evaluation index (HMEI) and communication degree (CD) optimal risk on the human to assess pollution in water of Habbaniyah Lake.

## 2. MATERIALS AND METHODS

## 2.1 Study area

Habbaniyah Lake (HL) is centrally situated in Iraq  $(33^{\circ}18'21''N 43^{\circ}26'14''E) 51$  m above sea level in height, as seen in Figure 1. Its depth is 9-13 meters, its length is 35 kilometers, its maximum width is 25 kilometers, and its storage capacity is 3.26 x10<sup>9</sup> cubic meters. The Euphrates River feeds the Habbaniyah Lake (HL) by the Warrar stream, The Majara Canal releases the water into Lake Razzazah, and the Theban Canal recirculates the water in the Euphrates River.



Figure 1. Location of the study area of HL

## 2.2 Sampling

In this study was selected thirty-three sites around the research area from October 2020 to May 2021, with water samples being determined using (Garmin Gps map 60 Csx). Water samples were collected using clean Polyethylene bottles (2 L) and take (15 ml) water sample. The process of digesting water samples was added (37% HCl) and (65% HNO<sub>3</sub>) to water samples in a flask size of (100 ml), filtered the samples

through Whatman paper (No. 42) to exclude unwanted objects (e.g., wood pieces, plants, and other particles) and keep at  $(4^{\circ}C)$  for more study. Then by using a hot plate at  $(200^{\circ}C)$  were heated samples to match the sample size (1 ml) with the addition of double water (24 ml) in the flask size (25 ml). The analysis was then carried out using an atomic absorption spectrophotometer (USA Phoenix-986).

Nine soil samples were taken from selected sits around the lake to quantify the HMs in soil, the samples were dried within the electrical oven at a 120°C for 3 hours and measuring one gram of every sample employing a sensitive balance. The process of digesting sediment samples was applied as water samples except that the sediment sample was one gram.

Sediments bed from HL were examined to measure the HMs concentrations, results show high concentrations of HMs in sediments bed. HMs in sediments bed were analysed as same soil samples [26].

# 2.3 Materials

(USA Phoenix-986) atomic absorption spectrophotometer (AAS) is the instrument that has been tested for heavy metal (HMs) analysis, Responsive balance, National Research Council-Canada ((NRCC), Regular (NASS-5) and Nitrate salt has been utilized to make some standard Ni, Cd, Cr, Fe, Co, Zn, Cu, and Pb solutions, using geographic information systems technology (ARC GIS.V 10.33) in research mapping and determination of water samples by using (GPS).

## 2.4 Evaluation pollution of heavy metal

To evaluate the portability of water and the effect of heavy metals on humans and their formulations dealing with relevant heavy metal features, the indices used for the study were developed: the Heavy Metal Evaluation Index (HEI), the Heavy Metal Pollution Index (HPI), and the Containment Degree (CD) [8, 16]. For calculating indices, a standard was used HPI and HEI calculation values (WHO guidelines) (Table 1).

 
 Table 1. Applied HMPI and HMEI estimation parameters and constants (according to WHO guidelines)

Metal	MCL*mg/l	Wi	Ii	Si	MAC
Co	0.05	20	0.004	0.11	0.05
Fe	0.2	5	0.2	0.3	0.2
Pb	0.0015	666.67	0.01	0.1	0.0015
Cr	0.05	20	0.05	0.1	0.05
Ni	0.02	50	0	0.02	0.02
Zn	5	0.2	0.2	5	5
Cd	0.003	300	0	0.005	0.003
Cu	1	1	0.05	1	1

Wi =Weightage (1/MAC), Si = Allowable standard in ppm, Ii = Highest allowed in ppm, MAC: Maximum allowable concentration.

#### 2.4.1 Contamination factor and contamination degree

In this analysis the contaminating CF element and the extent of contamination used to define water pollution. For the concept of pollution element, CF values are recommended [10]. Cf < 1 (low factor); 1 to Cf < 3 (moderate factor); 3 = Cf < 6 (significant factor of pollution); Cf =6 (very high contamination factor). Both emission causes were detected for the level of pollution (CD).CD < 6(low pollution degree); 6 = CD < 12 (moderate pollution degree); 12 = cd < 24 (significant pollution degree); cd = 24 (High contamination level suggesting significant anthropogenic contamination). The pollution degree (CD) and pollution Factor (Cf) are calculated from Eqns. (1) and (2):

$$CD = \sum_{i=1}^{n} Cfi \tag{1}$$

$$Cfi = \frac{c_{Ai}}{c_{Ni}} - 1 \tag{2}$$

The analytical value for ith is the Cai element, and The CNi in the ith section is the maximum allowable concentration [27].

2.4.2 Index of background enhancement (Indices calculation) (1) Heavy Metal Pollution Index (HMPI)

The HMPI index is one of the highest rating methodologies and a very valuable instrument that offers a cumulative impact on water quality by individual heavy metals [28]. The overall level for heavy metals in drinking water is shown [29]. This index shall be determined under the following Eqns. (3) and (4):

$$HMPI = \frac{\sum_{i=1}^{n} Wi.Qi}{\sum_{i=1}^{n} Wi}$$
(3)

$$Q_i = \sum_{i=1}^n \frac{Mi(-)Ii}{Si-Ii} \,\widehat{n} \,100 \tag{4}$$

where, Qi and Wi are the subs for ith and the weighting units for ith indices. n takes the number of variables into consideration. The heavy metal monitoring values of Mi, Ii, and Si are both the optimal and normal parameters of the ith individual. The sign (–) represents values number that overlook the algebraic symbol. Includes, low (under 100), high (over 100) threshold risk (equal 100). If the water content is higher than 100, the water cannot be used as drinking [28, 29].

(2) Heavy Metal Evaluation Index (HMEI)

The HMEI measures the water's total quality in terms of heavy metal content [8]. HMEI is an indicator of heavy metal water content in potable water [29]. HMEI index is determined as follows by Eq. (5):

$$HMEI = \sum_{i=1}^{n} \frac{Hc}{Hmac}$$
(5)

where, Hc and Hmac are the tracked and maximally allowable concentrations of the ith parameters. The HEI index classifications are low (less than 10), medium (between 10 and 20), and heavy (more than 20).

## 2.5 Statistical analysis

Multivariate procedure gives appropriate quantitative and qualitative data about the pollutant's origins [30]. Applying of a single Multivariate method alone to water quality data can yield highly important information on water quality [31, 32]. We apply two Multivariate methods in this study: Principal Component Analysis (PCA) and Cluster Analysis (CA).

PCA's goals are to extract underlying data from multivariable raw data as a set of uncorrelated (i.e., orthogonal) variables. PCA examines a data set containing observations characterized by a number of implied variables that are generally, inter-correlated.

These variables are known as principal components, singular vectors, eigenvectors, or loadings. Each unit also creates a list of scores based on its projection on the components.

The significance of every component is represented by the variance of its projection or by the ratio of the variance illustrated [33].

The vertical rotation was used so as to reduces the variables number of a high loading on every component and improves the understanding of relations and the potential origin of metals, lithogenic or anthropogenic [34, 35].

In the context of the sources of water contaminants, the multivariate statistic approach of a limited collection of data combined with additional information might give important insight about pollutant origins [36, 37].

Cluster analysis (CA) is one of predictive data analysis techniques that is used to solve classification issues. Its purpose is to group or cluster situations, data, or objects. The clusters should have high exterior (between-cluster) heterogeneity and high internal (within-cluster) homogeneity [38].

The most popular technique, hierarchical CA, begins with each issue in a new cluster and gradually links clusters together until only one cluster produces [39].

A dendrogram or binary tree is the best way to represent the results of applying cluster analysis. further depth about this approach by Salah et al. [40].

Pearson's correlation, PCA, and CA were performed applying IBM SPSS Statistic 26 software to investigate the heavy metals origin in water of HL.

## **3. RESULTS AND DISCUSSION**

## 3.1 Concentration of heavy metal in water

Table 2 objectively determines the amounts of heavy metals obtained from the HL in the area of research. Figures 2 (2a&2b), 3(3a&3b), 4(4a&4b) and 5(5a&5b) show the Spatial and temporal concentration during wet(a) and dry season(b) of (Pb), (Fe), (Cr) and (Co) respectively. In the dry order, the mean concentration of the metals analyzed in the water is as follows: Cr >Ni >Fe>Co>Pb>Zn=Cd=Cu and in the wet season Cr>Fe>Pb>Co>Ni=Zn=Cd=Cu. During the (dry and wet) seasons, the average concentration of Cr was 0.209 and 0.164 mg/l, respectively, above WHO, FAO, IQS, EPA and CCME drinking, irrigation, Iraqi Standards, Environmental Protection Agency, and aquatic life standards, respectively. During the dry and wet seasons, average levels of Fe inside the WHO, FAO, EPA, IQS and CCME were 0.035 and 0.037 mg/l respectively. Interestingly, the maximum Fe concentration was found at the W3 site (0.0885 mg/l) during the wet season. The average Co concentration in the wet (0.0153 mg/l) was less than in dry (0.024 mg/l), but it was also higher than the WHO, IQS, and CCME limits. The average values in Pb of water during dry and wet seasons was 0.0205 mg/l and 0.0206 mg/l respectively, above (WHO), (EPA) and (CCME) recommendations for marine life and below the allowable limits of Food, (FAO) and IQS guidelines (Table 2). The mean values of (Cd, Zn, and Cu) were non-consistent in all stations during wet and dry seasons due to the standard of (AAS), the device did not measure these metals due to very low concentrations of these metals were within the allowable limits of WHO, FAO, IQS, EPA, and CCME guidelines (Table 2). During the dry season Ni was above the CCME limit of aquatic organisms, an average of 0.07 mg/l, during dry time attributable to agricultural and human activity, the highest value of Ni was found in the site W3 (0.097 mg/l), exceeding

the standards of the WHO and CCME limits for drinking and living water and during the wet season the mean value of Ni was non-consistent. The values of Ni were recorded in Warrar stream during the dry season and not recorded in HL because a concentration was diluted with the river water and therefore concentration is non-consistent. The maximum values of the measured metals (Fe and Co) were recorded at S1 during the dry season this can be due to the lack of treatment stations for inlet water in Lake as well as this water from industrial facilities such as sand and gravel facilities that located near the lake, in additions auto laundry, fat and gasoline resulting from cars and river boats that leads to increases the concentration of heavy metals. Seasonal differences of metals in water were found, with dry showing higher amounts than wet [41]. Higher levels of Cr and Ni in the dry season and lower in the wet season were observed. The high level of Pb at S1 and W3 in water during wet and dry season might be due to industrial and agriculture waste and erosion sources, stations vary spatially and can be caused by the distribution of elements affecting many significant spatial variables, such as human population, Lake bank densities, hydrological conditions on the beds, dumping by local factories, and wastewater disposals [2].

Which has large amounts of fuel combustion lead in cars, was also caused by Concentrations of heavy metals analyzed between various the spillover of fuel from fishing vessels. Due to the heavy fuel consumption causes higher Pb levels in water bodies [42].

Results show that water of Habbaniyah Lake was highly polluted by HMs, therefore extra research done to know the HMs origin source. wastewater discharge points, Soil samples and sediments bed were taken from specified agricultural areas, and Habbaniyah Lake sediments bed.



Figure 2. Spatial and temporal concentration of Lead (Pb) during: (a) wet, and (b) dry season



Figure 3. Spatial and temporal concentration of Iron (Fe) during: (a) wet, and (b) dry season



Figure 4. Spatial and temporal concentration of Chromium (Cr) during: (a) wet, and (b) dry season



Figure 5. Spatial and temporal concentration of Cobalt (Co) during: (a) wet, and (b) dry season

Table 2. Heavy metals analytical value and guidelines (mg/l)

HMs	Average HMs Conc.±S.D.	D (a)	I (p)	AL (c)	IQS	EPA
Pb	$0.02 \pm 0.0045$	0.01	5	0.007	0.05	0.015
Fe	$0.036 \pm 0.018$	0.3	5	0.3	0.1	0.3
Cr	0.187 ±0.096	0.05	0.1	0.001	0.05	0.1
Cd	$0.000 \pm 0.000$	0.003	0.01	0.001	0.005	0.005
Ni	$0.035 \pm 0.009$	0.07	0.2	0.025	0.2	-
Со	$0.019 \pm 0.019$	0.01	0.1	0.002	0.01	-
Zn	$0.00\pm0.00$	3	2	0.005	5	5
Cu	$0.00\pm0.00$	2	0.2	0.005	1	1.3

D: Driking; I: Irrigation; AL: Aquatic lif ;(a): WHO (2011); (b): FAO (1994); (c): CCME (2007); IQS (2005) Iraqi Standard; EPA (2009) Environmental Protection Agency.

## 3.2 Wastewater discharge

There are three point source discharging directly the untreated wastewater into Warrar canal, Then pollutants transformed indirectly into Habbaniyah lake [43].

Tests were applied to measure HMs in wastewater show the wastewater polluted with heavy metals. Table 4 show the heavy metals concentration in wastewater discharged indirectly to Habbaniyah Lake.

#### 3.3 Heavy metal in sediments bed

Sediments beds from HL were analysed to measure the HMs concentrations. Results of HMS concentration in sediments bed of HL shown in Table 3.

#### 3.4 Heavy metals in soil

Nine soil samples were taken from selected sits around the lake to quantify the HMs in soil. Results of HMS concentration in soil of HL shown in Table 3.

#### 3.5 Assessment of metal pollution

Contamination degree (CD) has been used to measure metal pollution levels [16]. Table 5 shows the heavy metal distribution water quality indices spatially and temporally (CD, HMEI, and HMPI). Figures 6(6a&6b), 7(7a&7b) and 8 (8a&8b) show the spatial and temporal variation during the dry and wet season of (CD, HMEI, and HMPI) respectively. The average CD value of samples in the wet season is 20.89848, while this value decreases notably during dry to 17.37455, so the regions are classified considerable degree of contamination. The maximum pollution degree recorded in S1 during the wet and dry season were recorded in the wet season (53.29) and

the dry season (38.27), classified a pollution level that is very high because of agricultural and human activates and sewage without treatment. The lowest contamination degree recorded in S5 (5.77) classified a low degree of contamination and S28 (10.59) classified a moderate degree of contamination during wet and dry season respectively. Contamination of heavy metal in aquatic environment and heavy metal pollution have also been evaluated with the HMPI and HMEI. The average HMPI value during the wet season is (57.8248), while during the dry season (61.87697). The HMPI calculated shows values ranging from 16.56 and 101.56 in the wet season (mean 57.8248) and the dry season vary from 21.3 to 108.07 (mean 61.87697) according to the standard of HMPI that the water of the Lake is low contamination by heavy metal. The mean value of (HMEI) was 24.89772 and 22.37479 during the wet and dry seasons respectively, the mean value of (HMEI) increased during the wet season so this indicates high heavy metals level pollution and inappropriate for drinking water samples have been found in the research region. The maximum values of CD and HMEI were recorded in S1 during seasons due to use fertilizers and chemicals materials in agriculture that are added to the agricultural soil to improve soil quality and use the surface irrigation that effects to the water return to HL from this agricultural fields especially those irrigation projects are located along the Euphrates River, in addition to ground water, rural wastewater, Population growth that gets in the area, and industrial activities near these stations can be attributed to high values. The minimum values of CD and HMEI were recorded in S5 in the wet season while registered in S28 in dry season these regions classified had low contamination by heavy metals because there is no agricultural and industrial activity and far from the population assembly in this region. The maximum value of (HMPI) was recorded in S7 (101.56) because of industrial activity while in the dry season was recorded in S6 (108.07) due to human activity and sewage without treatment these areas were classified high polluted by heavy metals. The minimum value of (HMPI) recorded in S5 (16.56) during the wet season and S5 (21.3) during the dry season because this area is far from the population and agricultural activity. The HMPI values of this analysis show that water from the lake is low polluted, which based on classification of HMPI method (average HMPI 57.82 and 61.87 for wet and dry respectively) so that leads to low polluted by heavy metals.

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Table 3. Average Heavy metals analytical value in soil and sediments beds and guidelines

	W	astewater mg/L			Soil mg/Kg	Sediment	mg/Kg
Metals	Wastewater	IQS Discharge <sup>(a)</sup>	EPA	soil	USEPA guideline	Sediments bed mg/ kg	USEPA guideline
Pb	0.5756	0.05	0.015	1.337	40	47.354	40
Fe	3.7475	0.3	0.3	245	30	2152.971	30
Cr	0.6371	0.05	0.1	1.413	25	25.188	25
Ni	0.4523	0.1	-	4.47	16	18.002	16
Со	0.269	0.05	-	1.328	-	10.608	-

IQS (2005) Iraqi Standard, EPA (2009) Environmental Protection Agency, USEPA [1999]



Figure 6. Spatial and temporal variation of (CD) during: (a) wet, and (b) dry season



Figure 7. Spatial and temporal variation of (HMEI) during: (a) wet, and (b) dry season



Figure 8. Spatial and temporal variation of (HMPI) during: (a) wet, and (b) dry season

#### **3.6 Pollution Source**

The principal component analysis (PCA) was performed to investigate the relationship among the five metals analyzed. Table 4 shows the Pearson correlation coefficients results and the associated matrix developed in the study.

Table 4. PCA matrix of HMs during wet and dry seasons

Wet season (P < 0.05)							
	Cr	Pb	Fe	Со			
Cr	1.000						
Pb	0.416*	1.000					
Fe	-0.221	-0.251	1.000				
Co	-0.728**	-0.191	-0.319	1.000			
<b>Dry season</b> ( <b>P</b> < <b>0.05</b> )							
	Ni	Cr	Pb	Fe	Со		
Ni	1.000						
Cr	-0.220	1.000					
Pb	$0.408^{*}$	0.035	1.000				
Fe	0.482**	-0.284	0.315	1.000			
Co	-0.022	-0.569**	-0.247	0.026	1.000		
*. Correlation is significant at the 0.05 level (2-tailed).							

\*\*. Correlation is significant at the 0.01 level (2-tailed).

 Table 5. PCA (Rotated matrix) of HMs during wet and dry seasons

Wet season						
Element	Factor 1	Factor 2				
Cr	0.454	0.171				
Pb	0.205	0.426				
Fe	0.142	-0.678				
Со	-0.517	0.259				
Eigen value	1.934	1.322				
Variance explained %	48.354	33.049				
Cumulative variance %	48.354	81.404				
Dry season						
Element	Factor 1	Factor 2				
Ni	0.437	0.034				
Cr	-0.103	-0.51				
Pb	0.4	-0.212				
Fe	0.405	0.105				
Со	-0.104	0.534				
Eigen value	1.907	1.627				
Variance explained %	38.143	32.535				
Cumulative variance %	38.143	70.678				

The principal components with eigenvalues larger than 1 were extracted with loadings rotated for the maximum variance. Two factors (F1 and F2) were extracted, accounting for wet season were 48.354% and 33.049% respectively, of the total variance of 81.404% as shown in Table 5. (F1 and F2) were extracted, accounting for dry season were 38.143% and 32.535% respectively, of the total variance of 70.678%. Table 5 provides numerical representation for the relations between the HMs and the two factors, during wet season, results show moderately connection between Cr and factor 1, and also moderate relation between Pb and factor 2.

In addition, during dry season, results show moderately connection among Ni, Pb and Fe with factor 1, and also moderate relation between Co and factor 2. This means HMs originated from multi sources (lithogenic and anthropogenic). Results confirmed the estimation provided by Table 4, that the relation between HMs are either insignificance or moderately, and the optimal origin from the factor are moderately, therefore the major pollutant source may be contributed from anthropogenic origin, which mostly due to the wastewater point sources on Warrar stream, and the minor source is lithogenic. Two components (F1 and F2) were with an eigenvalue greater than 1, which points the exist of multi sources of HMs in HL.

#### 4. CONCLUSION

In the study, water quality of Habbaniyah lake (HL) was analyzed and distribution characteristics of heavy metals (Cd, Cu, Pb, Cr, Zn, Ni, Fe, Co) pollution during two seasons (dry and wet). The concentrations of heavy metals during dry season were a greater than wet season, there was simple difference between the two seasons, The average heavy metal concentration in water over the dry is as follows: Cr>Ni>Fe>Co>Pb>Zn=Cd=Cu while follows Cr>Fe>Pb>Co>Ni=Zn= Cd=Cu over wet seasons.

The mean levels of some HMs concentrations were greater than the safe prescribed by WHO, CCME, and EPA. HMPI, (HMEI), and (CD) were because in dry season agricultural crops need large amounts of water and use of surface irrigation style, in addition to the fertilizers and chemicals materials in agriculture that are added to the agricultural soil to improve soil quality that produce the heavy metals that effect on the lake and industrial activities, human population, sewage without treatment and tourist activity during the dry season. In all stations the average (Cd, Zn and Cu) values during the seasons were inconsistent due to the standard of (AAS) the device did not feel these metals due to lack of concentrations of these metals. High concentrations of average HMs in HL water are due to: 1) HL represents the final discharge of many effluent point sources from agricultural and residential areas without treatment; 2) The bad water quality of Euphrates River at upstream; 3) The surrounding area of the HL has been subjected to military activities and bombings in previous years; we believe that military actions may be a significant source of elevated heavy metals such as (Ni, Cu, Zn, Pb) in the studied region; 4) using inappropriate fishing methods such as (toxic chemicals and explosives); 5) Industrial activates and population growth. All these reasons lead to the increase of the concentrations of heavy metals in water.

Finally, results indicate that HL is highly contaminated with HMs according to national and international guidelines (CD), (HMPI) and (HMEI) indices pointed that HL water quality was bad.

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