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Combined Effect of Heat Source and Pressure in an Unsteady MHD Flow with Chemical Reaction Through an Infinite Oscillating Vertical Porous Surface



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pressure gradient, heat source. magnetohydrodynamic (MHD), porous medium, chemical reaction parameter

ABSTRACT

In this study, heat source, pressure gradient, and chemical reaction are investigated in the analytical solution of an unsteady magnetohydrodynamic fluid with heat and mass transfer along a vertical oscillatory surface under the porous medium. This study also investigates the temperature and concentration along with velocity, closer to the surface over time t>0. The Laplace transformation procedure has been implemented for resolving the governing terms of the flow, yielding solutions for temperature, concentration, and velocity. Significant factors that affect velocity, temperature, and concentration—including the Schmidt number, thermal Grashof number, and solutal Grashof number are analysed and results are illustrated graphically using MATLAB. The results obtained here are consistent with the previously published work. From the current investigation, it is reported that velocity decreases with a rise in Hartmann's number (M). As the chemical reaction Kr rises, the velocity has a retarding effect, whereas the concentration increases, and while the heat source (Qs) increases corresponding velocity and temperature both are increasing. As pressure gradient P increases, velocity decreases gradually.

1. INTRODUCTION

An examination of free convection and heat transfer in permeable flows responsive to magnetism has garnered substantial attention due to its extensive scientific and technological implications. Film vaporization in combustion systems, surface ablation procedures, and cooling techniques for rocket boosters and re-entry vehicles are some examples of these uses. Furthermore, flows in permeable materials are essential in numerous engineering and geophysical fields, including filtration and purification in chemical processes, the examination of groundwater circulation in agricultural engineering, and the assessment of fluid dynamics in petroleum reservoirs encompassing oil, natural gas, and water. Because of this, many studies have focused magnetohydrodynamic (MHD) free convection, weight transfer on permeable surfaces with diverse shapes.

Unsteady MHD flow in porous medium over an oscillating vertical plate has been thoroughly examined by Krishna et al. [1], who analysed heat and mass transfer effects in such a configuration. Krishna and Chamkha [2] and Krishna [3] explored the impacts of hall and ion slip currents in rotating nanofluid and Jeffreys fluid flow respectively. The inclusion of ion slip and porous effects in a nanofluid setting was extended by Krishna and Chamkha [4], while heat source and chemical reaction effects in viscoelastic MHD flows were discussed by Krishna and Jyothi [5]. These studies provide fundamental insights into how electromagnetic and reactive phenomena influence transport processes.

Khanam and Ahmed [6] and Anand Kumar et al. [7] mined the interplay of thermal radiation, mass diffusion, and chemical kinetics in porous channels, showing how reaction rates can drastically modify concentration profiles. The incorporation of thermal radiation, often modeled using the Rosseland approximation, has been addressed by Merkin et al. [8] and Ahammad et al. [9]. Hari Babu [10] focused on chemically reactive Casson and viscous fluids in porous setups, illustrating how increasing the reaction parameter leads to significant attenuation of species concentration due to stronger destructive reactions. Furthermore, MHD convection in porous media has been a focal point in works like Yedhiri et al. [11], Ayub et al. [12], and Raju et al. [13] have examined the effects of thermal radiation and internal heat generation on the MHD flow of Casson fluids within a rotating porous medium. Their study focuses on flow induced by an inclined, oscillating vertical plate an arrangement that closely represents practical applications like oscillatory pumps and industrial separators. Chandra Sekhar and Vijaya Kumar [14] discussed how increasing the chemical reaction parameter leads to significant attenuation of species concentration. Islam and Hossain [15], who demonstrated that an increase in the radiation parameter leads to a decrease in temperature gradients, thus influencing the buoyancy-driven flow behaviour. These thermal effects become more complex when heat generation or absorption is introduced simultaneously, as shown by Goud and Reddy [16], Kavitha and Angel [17]. Gokulavani et al. [18], who studied the effects of permeability, Hall currents, and nanofluid properties on boundary layer characteristics. The role of the porous matrix in regulating momentum and thermal dispersion is crucial for applications such as underground thermal storage, insulation materials, and biological transport processes. Tetbirt et al. [19] conducted a comprehensive numerical evaluation of MHD mixed-convection and the heat mobility involving immiscible micropolar along with nanofluids within a vertical channel under a transverse magnetic field. The authors incorporated critical physical mechanisms, including Brownian motion, thermophoresis, magnetic forces, and microrotation, illustrating their combined effects on velocity, temperature, and nanoparticle concentration profiles.

Bormudoi and Ahmed [20] carried out a seminal investigation into unsteady MHD free-convective flow over a vertical plate subjected to parabolic ramped temperature and concentration boundary conditions. In addition, their model incorporates both the Soret (thermal diffusion) effect and the presence of an internal heat sink. Employing the Rosseland approximation for radiation and the Boussinesq assumption for buoyancy, they solved the governing non-linear partial differential equations using similarity transformations coupled with analytical techniques. Their results demonstrate notable transient behavior in velocity, temperature, and concentration fields, with ramped boundary conditions inducing timedependent acceleration and deceleration of flow. Sakthivel and Sivakami [21], discussed an analytical solution of unsteady MHD immiscible fluid with chemical reaction in a porous medium under heat source.

Patoliya and Gohil [22] studied the chemical reaction on MHD Casson fluid flow past an oscillating plate in porous medium with hall current effects, analytical and numerical methods are used to analyze the fluid behavior. More recently, Kesavarani and Lakshmi Priya [23] studied the unsteady MHD oscillatory flow in a two-phase channel, focusing on the influence of the Soret effect (thermal diffusion) and oscillation-induced mixing. Their results highlight the significant impact of Soret-driven mass transport on the concentration and velocity profiles, particularly under varying thermal conditions. Raptis and Perdikis [24] presented one of the earliest analytical treatments of unsteady free convection in a highly porous medium under the influence of thermal radiation. Their findings demonstrated that radiation enhances the velocity and temperature fields, particularly in media with strong radiative absorption.

Balarabe et al. [25] analyzed the MHD heat and mass transfer characteristics of a second-grade fluid in the presence of thermal radiation. Their study emphasized the role of viscoelasticity and radiation in altering flow resistance and thermal distribution. Complementing this, Sinha and Choudhury [26] investigated the behavior for the flow of the MHD fluid past a vertical permeability plate under the influence of a heat source, providing insights into how internal heat generation affects both the thermal and momentum boundary layers.

While the previous studies haven't systematically examined Pressure and Heat Source in an Unsteady MHD Flow. This study examines it, along with the effects of permeability, radiative heat transfer, and chemical reaction of Newtonian fluid across an oscillating, vertically, infinite, permeable plate. that is located inside a highly saturated porous medium, with changing mass diffusion.

2. CONSTRUCTION AND RESOLUTION OF THE PROBLEM

Figure 1 illustrates the unsteady MHD laminar boundary flow of a viscous, Newtonian, incompressible fluid in an immeasurable vertical plate moving upwards through a porous layer. Coordinate system is defined such a way that the x-axis runs along the vertical direction of the plate's motion, while zaxis is normal to the plate's surface. Initially, both the fluid and the plate are at a uniform temperature T_{∞} and solute concentration \mathcal{C}_{∞} , maintaining thermal and concentration equilibrium. At time t>0, the plate begins to move in its own plane with a velocity of $U_0e^{i\omega t}$. Simultaneously, the surface temperature of the plate increases with time. A uniform magnetic field of strength B_0 is applied transversely to the plate. Due to the assumption of a low magnetic Reynalds number, the induced magnetic field is very small. The chemical reaction is assumed to be first-order, where the reaction rate is proportional to the deviation of concentration from its ambient value. The fluid is assumed to be gray, nonscattering, and capable of absorbing and emitting thermal radiation without any scattering effects. The Boussinesq approximation is applied, allowing density variations only in the buoyancy term of the momentum equation [1].

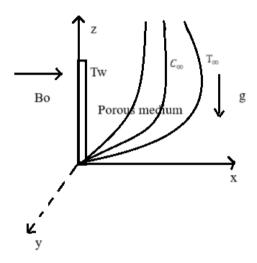


Figure 1. The problem's flow geometry

Then by Boussinesq approximation, the flow is ruled by the following:

$$\frac{\partial u^{I}}{\partial t} = v \frac{\partial^{2} u^{I}}{\partial z^{2}} - \left(\frac{\sigma B_{0}^{2}}{\rho} + \frac{v}{k}\right) u^{I} + g\beta (T^{I} - T_{\infty}^{I}) + g\beta (C^{I} - C_{\infty}^{I}) - \frac{\partial P}{\partial z}$$
(1)

$$\frac{\partial v^I}{\partial t} = v \frac{\partial^2 u^I}{\partial z^2} - \left(\frac{\sigma B_0^2}{\rho} + \frac{v}{k}\right) v^I \tag{2}$$

$$\rho C_{p} \frac{\partial T^{I}}{\partial t} = k_{1} \frac{\partial^{2} T^{I}}{\partial z^{2}} - \frac{\partial q r^{2}}{\partial z} + Q_{s} (T^{I} - T_{\infty}^{I})$$
 (3)

$$\frac{\partial c^I}{\partial t} = D \frac{\partial^2 c^I}{\partial z^2} - Kr(C^I - C_{\infty}^I)$$
 (4)

$$u^{I} = 0, v^{I} = 0, T^{I} = T_{\infty}^{I}, C^{I} = C_{\infty}^{I}, t \leq 0;$$

$$\forall z u^{I} = U_{0} e^{i\omega t}, v^{I} = 0, T^{I} = T_{\infty}^{I} + (T_{w}^{I} - T_{\infty}^{I})$$
At $C^{I} = C_{\infty}^{I} + (C_{w}^{I} - C_{\infty}^{I})A't$, for $t > 0$; $z = 0$

$$u^{I} \to 0, v^{I} \to 0, T^{I} \to T_{\infty}^{I}, C^{I} \to C_{\infty}^{I}, \text{ for } t > 0$$
as $z \to \infty$

Eqs. (1) and (2), let $q^I = u^I + iv^I$, it is obtained as:

$$\frac{\partial q^{I}}{\partial t} = v \frac{\partial^{2} q^{I}}{\partial z^{2}} - \left(\frac{\sigma B_{0}^{2}}{\rho} + \frac{v}{k}\right) q^{I} + g\beta (T^{I} - T_{\infty}^{I}) + g\beta (C^{I} - C_{\infty}^{I})$$
(6)

When considering a gray, optically dense gas, the local radiative absorption is specified as:

$$\frac{\partial q^I}{\partial z} = -4a\sigma(T_{\infty}^{I4} - T^{I4}) \tag{7}$$

Raptis and Perdikis [24] consider that the flow variations in the temperature are limited enough. Let's assume that the temperature variations along the flow route are sufficiently minimal that T^{I4} within the free stream thermal T^{I}_{∞} using Taylor's series without accounting for higher-level variations. This yields the subsequent approximated performance:

$$T^{I4} \cong 4T_{\infty}^{I3}T^{I} - 3T_{\infty}^{I4} \tag{8}$$

$$\rho C_{\rm p} \frac{\partial T^I}{\partial t} = k_1 \frac{\partial^2 T^I}{\partial z^2} - 16a\sigma T_{\infty}^3 (T^I - T_{\infty}^I)$$
 (9)

The nondimensional quantities:

$$\begin{split} z'' &= \frac{\omega_0 z}{v}, \frac{\partial q^l}{\partial z} = \frac{q'}{U_0}, \theta^l = \frac{T' - T_\infty}{T_w - T_\infty}, \Pr = \frac{\rho v C_p}{k_1}, K = \frac{U_0^2 k}{v^2}, \\ P &= \frac{-h^2}{\mu u^l} \left(\frac{\partial P}{\partial z} \right), Sc = \frac{v}{D}, \phi^l = \frac{C - C_\infty}{C_w - C_\infty}, M^2 = \frac{\sigma v B_0^2}{\rho U_0^2}, R = \\ \frac{16 a v \sigma T_\infty^3}{k_1 U_0^2}, Gm &= \frac{v g \beta (C_w - C_\infty)}{U_0^3}, Gr = \frac{v g \beta (T_w - T_\infty)}{U_0^3}, Kr = \frac{v K_1}{v_w^2}, \\ A &= \frac{U_0^2}{v}, t'' = \frac{U_0^2 t}{v}, N = M^2 + \frac{1}{K}, Q = \frac{Q s v}{\rho c_n U_0} \end{split}$$

By the transformation of Eq. (10), the nondimensional form of Eqs. (1), (4), and (9) are:

$$\frac{\partial \mathbf{q}^{\mathrm{I}}}{\partial \mathbf{r}} = \frac{\partial^{2} \mathbf{q}^{\mathrm{I}}}{\partial z^{2}} - \mathbf{N} \mathbf{q}^{\mathrm{I}} + \mathbf{G} \mathbf{r} \boldsymbol{\theta}^{\mathrm{I}} + \mathbf{G} \mathbf{m} \boldsymbol{\phi}^{\mathrm{I}} - \mathbf{P}$$
 (11)

$$\frac{\partial \theta^I}{\partial t} = \frac{1}{\Pr} \frac{\partial^2 \theta^I}{\partial z^2} - \frac{R}{\Pr} \theta^I + Q \theta^I \tag{12}$$

$$\frac{\partial \phi^{\mathrm{I}}}{\partial t} = \frac{1}{S_{\mathrm{C}}} \frac{\partial^{2} \phi^{\mathrm{I}}}{\partial z^{2}} - \mathrm{Kr} \phi^{\mathrm{I}}$$
 (13)

Initial boundary conditions [1]:

$$q^{I} = 0, \theta^{I} = 0, \phi^{I} = 0, t \le 0; \forall z$$

 $q^{I} = e^{i\omega t}, \theta^{I} = t, \phi^{I} = t, t > 0; \text{ at } z = 0$ (14)
 $q^{I} \to 0, \theta^{I} \to 0, \phi^{I} \to 0, t > 0; \text{ as } z \to \infty.$

The solution of concentration, temperature, and velocity distributions is obtained by mathematically solving the uncertain coupled differential equations with partial solutions. Eqs. (11)-(13) and the resulting boundary conditions (Eq. (14)) use the Laplace transformation approach, considering Eq. (13) and taking Laplace on both sides.

Using Laplace expansion:

$$L\left[\frac{\partial \phi^{I}}{\partial t}\right] = \frac{1}{Sc}L\left[\frac{\partial^{2} \phi^{I}}{\partial z^{2}}\right] - KrL[\phi^{I}]$$
 (15)

$$\frac{1}{Sc} \frac{d^2 \overline{\phi^I}}{dz^2} - Kr \overline{\phi^I} - S \overline{\phi^I} = 0$$
 (16)

Then we get:

$$\bar{\phi}_{(z,s)} = C_1 e^{-(\sqrt{Kr \cdot Sc + s \cdot Sc})z} + C_2 e^{(\sqrt{Kr \cdot Sc + s \cdot Sc})z}$$
 (17)

Then obtain C_1 and C_2 value using boundary conditions, $C_1 = \frac{1}{s^2}$ and $C_2 = 0$. Substituting C_1 , C_2 in Eq. (17).

 $\overline{\varphi}_{(z,s)} = \frac{e^{-(\sqrt{Kr\cdot Sc+s\cdot Sc})z}}{s^2} \ , \quad \text{now taking inverse Laplace}$ transform we get Eq. (18).

$$\phi^{I} = \left(\left(\frac{t}{2} + \frac{Z\sqrt{Sc}}{4\sqrt{Kr}} \right) \exp\left(Z\sqrt{Kr \cdot Sc} \right) \operatorname{erfc}(\eta\sqrt{Sc} + \sqrt{Kr \cdot t}) \right) + \left(\left(\frac{t}{2} - \frac{Z\sqrt{Sc}}{4\sqrt{Kr}} \right) \exp\left(-Z\sqrt{Kr \cdot Sc} \right) \operatorname{erfc}(\eta\sqrt{Sc} - \sqrt{Kr \cdot t}) \right)$$
(18)

$$\theta^{I} = \left(\left(\frac{t}{2} + \frac{z\sqrt{Pr}}{4\sqrt{B}} \right) \exp\left(Z\sqrt{B \cdot Pr} \right) \operatorname{erfc}(\eta\sqrt{Pr} + \sqrt{B \cdot t}) \right) + \left(\left(\frac{t}{2} - \frac{z\sqrt{Pr}}{4\sqrt{B}} \right) \exp\left(-Z\sqrt{B \cdot Pr} \right) \operatorname{erfc}(\eta\sqrt{Pr} - \sqrt{B \cdot t}) \right)$$
(19)

$$Q^{I} = \frac{1}{2} \exp\left(\frac{1+i\omega}{N}\right) t \left\{ \exp\left(-Z\sqrt{(1+i\omega)}\right) \operatorname{erfc}(\eta\sqrt{N} - \sqrt{\frac{1+i\omega}{N}}) t\right) + \exp\left(Z\sqrt{(1+i\omega)}\right) \operatorname{erfc}\left(\eta\sqrt{N} + \sqrt{\frac{1+i\omega}{N}}\right) t\right) \right\}$$

$$+ \frac{D}{k_{1}^{2}} \left\{ \left(\frac{1}{2} \exp\left(\frac{K_{2}}{P_{P}}\right) t\right) \left[\exp\left(-Z\sqrt{K_{2}}\right) \operatorname{erfc}\left(\eta\sqrt{P_{P}} - \sqrt{\frac{K_{2}}{P_{P}}}\right) t\right) + \exp\left(Z\sqrt{K_{2}}\right) \operatorname{erfc}\left(\eta\sqrt{P_{P}} + \sqrt{\frac{K_{2}}{P_{P}}}\right) t\right) \right\}$$

$$+ \sqrt{k} \left\{ \left(\frac{t}{2} + \frac{ZPr}{4\sqrt{K_{2}}}\right) \left[\exp\left(Z\sqrt{K_{2}}\right) \operatorname{erfc}\left(\frac{Z\sqrt{P_{P}}}{2\sqrt{t}} + \sqrt{\frac{K_{2}}{P_{P}}}\right) t\right) \right\}$$

$$+ \exp\left(\frac{t}{2}\right) \left\{ \left(\frac{1}{2} \exp\left(\frac{K_{2}}{P_{P}} + K_{1}\right) t\right) \left[\exp\left(-Z\sqrt{\frac{K_{2}}{P_{P}}} + K_{1}\right) \operatorname{Prerfc}\left(\eta\sqrt{P_{P}}\right) \right] \right\}$$

$$+ \frac{D}{k_{1}^{2}} \left\{ \left(\frac{1}{2} \exp\left(\frac{K_{2}}{P_{P}} + K_{1}\right) t\right) \left[\exp\left(-Z\sqrt{\frac{K_{2}}{P_{P}}} + K_{1}\right) \operatorname{Prerfc}\left(\eta\sqrt{P_{P}}\right) \right] \right\}$$

$$+ \exp\left(Z\sqrt{\frac{K_{2}}{P_{P}}} + K_{1}\right) t\right) \left\{ 20\right\}$$

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$$+ \exp\left(Z\sqrt{\frac{K_{2}}{P_{P}}} + K_{1}\right) t\right) \left\{ 20\right\}$$

$$+ \frac{F}{K_{3}^{2}} \left\{ \left(\frac{t}{2} + \frac{Z}{4}\sqrt{\frac{S_{C}}{K_{P}}}\right) \left[\exp\left(Z\sqrt{S_{C}} \cdot K_{P}\right) \operatorname{erfc}\left(\eta\sqrt{S_{C}} + \sqrt{K_{F}} \cdot t\right) \right] \right\}$$

$$+ \frac{F}{K_{3}^{2}} \left\{ \frac{1}{2} \exp\left(Kr - K_{2}\right) t \left[\exp\left(-Z\sqrt{\frac{K_{F}}{K_{F}}} - K_{3}\right) \operatorname{S_{C}}\right) \operatorname{erfc}\left(\eta\sqrt{S_{C}} - \sqrt{\frac{K_{C}}{K_{F}}} - \sqrt{\frac{K_{C}}{K_{C}}}\right) \right\}$$

$$+ \exp\left(Z\sqrt{\frac{K_{C}}{K_{C}}} - \sqrt{\frac{K_{C}}{K_{C}}}\right) \exp\left(-Z\sqrt{\frac{K_{C}}{K_{C}}} - \sqrt{\frac{K_{C}}{K_{C}}}\right) + \exp\left(2\sqrt{\frac{K_{C}}{K_{C}}} - \sqrt{\frac{K_{C}}{K_{C}}}\right) \exp\left(-Z\sqrt{\frac{K_{C}}{K_{C}}} - \sqrt{\frac{K_{C}}{K_{C}}}$$

 $+\sqrt{(Kr-K_3)t}$

3. RESULT AND DISCUSSION

The impact of the non-coupling parameters with radiation parameter (R), dimensionless time (t), and permeable parameter (K) along with Hartmann number (M) and Grashof number (Gr), on the velocity component velocity $q^{\rm I}$, temperature $\theta^{\rm I}$, and concentration $\varphi^{\rm I}$ has been numerically assessed. It explores the concentration, velocity, and temperature identities by altering variables by keeping the parameters fixed as the Grashof number (Gr=3), Gm=3>0, Prandtl number (Pr=0.71), Schmidt volume (Sc=0.22), Heat source (Qs=0.05), K=2, P=0.2, and M=0.5, the profile is plotted by varying one parameter at a time while holding the others constant. We employed MATLAB to collect and analyze the data.

According to Figure 2, the graphic illustrates that concentration diminishes as the Schmidt number (Sc) increases. Physically, this is accurate, as an increase in Sc results in a reduction of molecular diffusivity, thus leading to reduced concentrations at the edge of the layer. Consequently, low Sc values elevate the species concentration, while high Sc values diminish it. Figure 3 illustrates that how chemical reaction affects the concentration profile if destructive chemical reaction increasing the reaction rate leads to lower concentration, because more solute is consumed by the reaction. The concentration is observed to increase during time t as shown in Figure 4. Figure 5 illustrates the transient temperature distribution with respect to z. The temperature attains its extreme value gradually decreases to zero as the distance increase. The temperature magnitude is higher for air (Pr=0.71) compared to water (Pr=7), which is particularly attributed to the drop in thermal conductivity with increasing Prandtl number. Figure 6 illustrates the radiation variable, the radiation parameter (R) of the temperature profiles shows a minor decrease in association with the thermal conductivity ratio, which increases with the magnitude of the radiation factor this is due to the radiating effect, which causes the gas to transmit heat onto the surface. The gas or liquid gets more viscous as a result of its temperature dropping. Figure 7 illustrates how Qs affects temperature profiles. It is evident that due to the added heat energy impacts the temperature profile proportionally.

Figure 8 shows the influence of the temperature Grashof number (Gr) with the velocity gradient. The study implicates the increase in Gr correlates with a rise in acceleration. In Figure 9, the mass Grashof number (Gm) dictates the relationship between buoyant and viscous dynamical forces. As the buoyancy force of a species intensifies, fluid velocity diminishes, resulting in a more pronounced peak value. The range of velocity attains its maximum near the plate before progressively diminishing to the empty stream frequency, on the field of motion. The velocity increases in association with increasing Solutal Grashof quantity ranges.

As Pr increases, velocity component decreases (Figures 10-11). Moreover, the acceleration diminishes and approaches its smallest value near the plate. The velocity components diminish as the Schmidt ratio (Sc) increases. Figure 12 illustrates an increase in the radiation parameter (R) velocity component as it decreases, because radiation damps the temperature field, thereby weakening buoyancy driven flow which reduces the overall fluid acceleration.

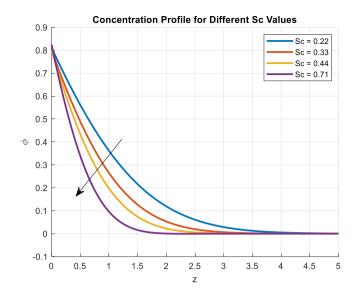


Figure 2. Influence of the Schmidt number (Sc) for concentration

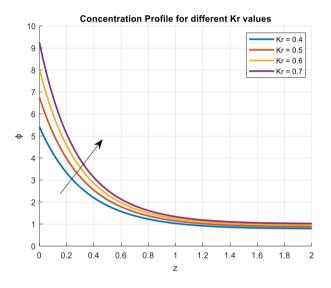


Figure 3. Influence of the chemical reaction (Kr) for concentration

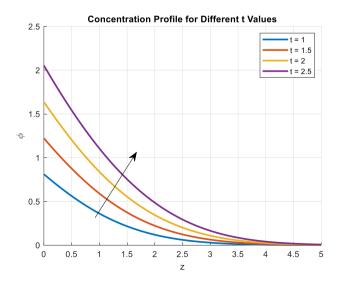


Figure 4. Influence of time for the concentration profile

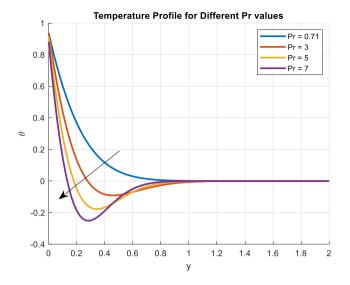


Figure 5. Influence of Prandtl number (Pr) for the temperature profile

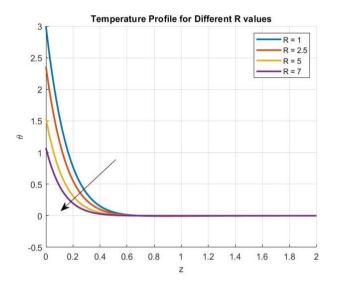


Figure 6. Consequence of Radiation parameter (R) for the temperature

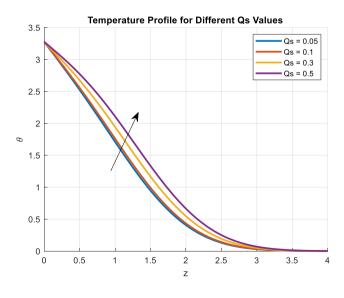


Figure 7. Impression of Heat source (Qs) for the temperature profile

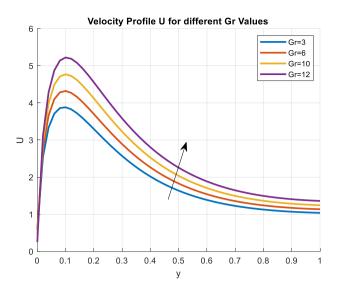


Figure 8. Influence of the Grashof number (Gr) for velocity

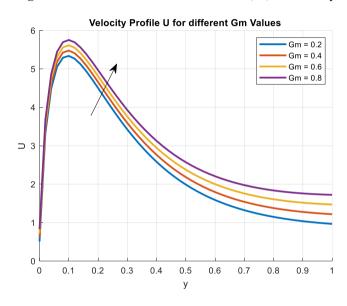


Figure 9. Effect of the Grashof number (Gm) for velocity

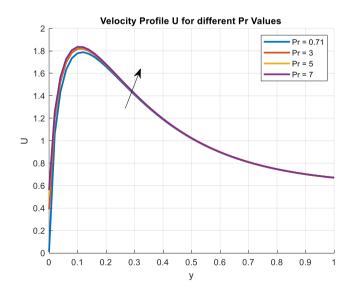


Figure 10. Effect of the Prandtl number (Pr) for the velocity

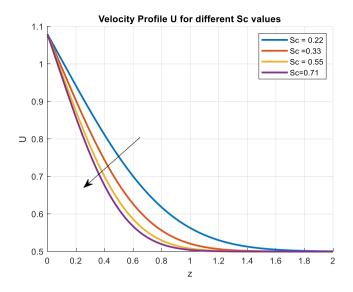


Figure 11. Influence of Schmidt number (Sc) for the velocity

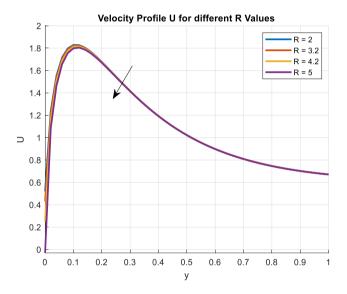


Figure 12. Influence of Radiation parameter (R) for the velocity

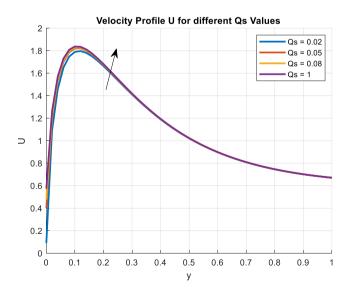


Figure 13. Effect of Heat source (Qs) for the velocity profile

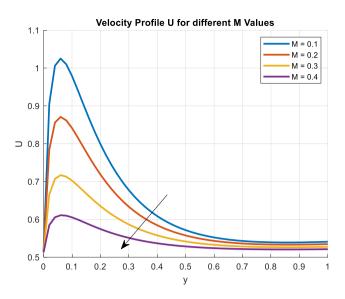


Figure 14. Influence of Magnetic parameter (M) for the velocity

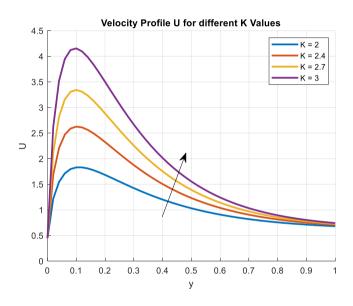


Figure 15. Effect of Porous parameter (K) for the velocity

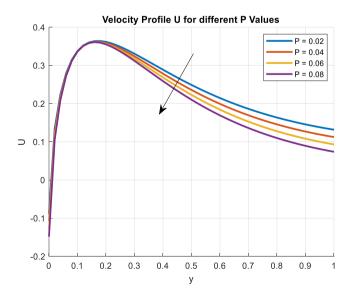


Figure 16. Effect of Pressure (P) for the velocity profile

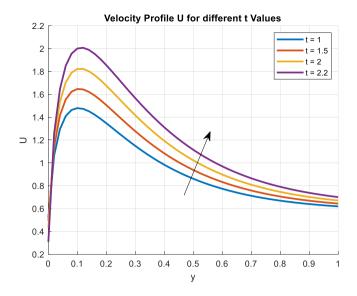


Figure 17. Influence of time parameter (t) for the velocity profile

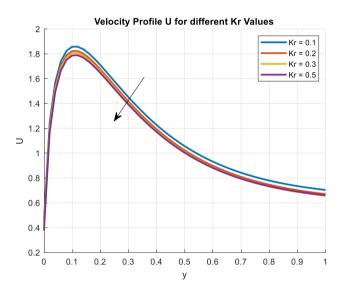


Figure 18. Influence of chemical reaction (Kr) for the velocity

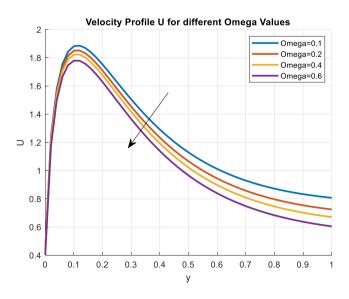


Figure 19. Effect of oscillating parameter (ω) for the velocity profile

In Figure 13, on Qs for velocity profiles this graphic illustrates whether the generation of heat enhances the buoyancy force, resulting in an increased flow rate and subsequently elevated acceleration profiles. The velocity features in Figure 14 exhibit a retardation as the magnetic force amplitude (Hartmann number M) increases. It is evident that acceleration decreases at all sites when the regional magnetic factor M increases. The imposition of a longitudinal magnetic field on an electrically conductor fluid generates a resistance force known as the Lorentz force.

Figure 15 demonstrates how the velocity components vary as the permeability parameter K changes. As the permeability parameter K increases, the velocity components get higher. The flow velocity diminishes when a porous material is present because it raises the resistance to flow. The velocity's decline with decreasing K exhibits this behaviour. Figure 16 indicates the way pressure affects the acceleration profile; as the pressure level rises, so does the velocity profile decreases. As pressure increases along the direction of flow, the velocity profile decreases due to the opposing pressure force acting against the fluid motion. Figure 17 shows an increase in velocity as time t passes. Figure 18 illustrates how the velocity fields are affected by the chemical reaction parameter Kr. These numbers show that the velocity fields decrease as Kr rises due to enhanced resistance in the flow field caused by the interaction of chemical species. According to physical principles, the flow field naturally decreases as Kr increases. Lastly, Figure 19 reveals that velocity increases as oscillation frequency ω increases. Table 1 illustrates the comparison result of the previous work carried out by Krishna et al. [1] and the present study. Where in addition with heat source leads to a slight but consistent increase in temperature profile. Which correlates with the existing system.

Table 1. Comparison of results for temperature profile

Pr	Previous Result [1] Laplace Transform Technique	Present Results Laplace Transform Technique, Adding Qs
0.71	0.0711681	0.0711772
3	0.0025271	0.0025401
7	0.0000186	0.0000192

4. CONCLUSIONS

Using a flux model, we have reproduced the effects of heat radiation in very dense gases. The following conclusions are obtained by applying the Laplace transform methodology with appropriate bounds to the analytical results of the governing equations:

- (1). The velocity frequently diminishes with a rise in Hartmann's number M. and oscillation ω decreases in the velocity flow.
- (2). When Grashof number, Gr and Gc, increase velocity gradually increases. Which increases buoyancy forces, and fluid flow rises.
- (3). As time increases, the velocity, temperature, and concentration are increasingly enhanced.
- (4). The acceleration and temperature fluctuated with variations in the free convection-radiation factor.
- (5). A rise in Sc along with time results in a progressive acceleration of the energy transfer coefficient.
- (6). As chemical reaction Kr rises the velocity profile reduced, and concentration increases.

(7). As heat source Qs moves up as the velocity increase and temperature decrease.

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NOMENCLATURE

u,v Velocity componentst Dimensionless timeP Pressure source

- B_0 Coefficient of electromagnetic field
- R Thermal Radiation parameter
- θ^I Dimensionless temperatureK Permeability parameter
- *k* Permeability of porous medium
- v Kinematic viscosity
- μ Fluid viscosity
- g Acceleration due to gravity
- *q^I* Velocity parameter
- ϕ^I Dimensionless concentration
- T_W Fluid temperature at plate
- C_W Fluid concentration at plate
- Gr Thermal Grashof number
- Gc Mass Grashof number
- M Hartmann number
- Pr Prandtl number
- Sc Schmidt number
- α Viscosity ratio
- β Coefficient of heat transfer
- ρ Density
- σ Electrical conductivity
- ω Frequency parameter
- a Spectral mean absorption coefficient of the medium
- ∞ free stream conditions
- U_0 Dimensionless plate velocity
- D Chemical molecular diffusivity
- C_p Heat at constant pressure
- Kr Chemical reaction parameter
- Qs Heat source parameter