

Optimization of Hot-pressing Technique for Genuine Full-grain Leather Made of Natural Rubber Latex

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https://doi.org/10.18280/acsm.430305 ABSTRACT

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Keywords:

natural rubber latex, genuine full-grain leather (GFGL), hot-pressing technique, physical-mechanical properties Currently, the waste shavings are either directly recycled to produce regenerated leather or used to extract collagen. Both methods are complex and unfriendly to the environment. This paper aims to recycle waste chrome shavings in a clean, simple and low-consumption manner. For this purpose, the hot-pressing method was selected to prepare genuine full-grain leather (GFGL) plates. Then, the author probed into the influencing factors and their impacts of hot-pressing products of prefabricated plates. Through the research, the optimal hot-pressing conditions and formula for GFGL plates made of natural rubber latex were determined as: (1) latex ratio 15 %, water ratio 2:1, 110 °C, 6min and 5MPa; (2) latex ratio 21 %, water ratio 2.5:1, 100 °C, 4min and 5MPa. Note that the latex ratio refers to the weight ratio of water to latex.

1. INTRODUCTION

The statistics of the UK Leather Trade Federation show that only 20% of tanning materials are converted into leather. Even if the lining leather is considered, the utilization rate only increases to 31.5% [1, 2]. The other materials become solid wastes through such operations as splitting, shaving, trimming and buffing. Each ton of wet-salted hides can only produce 200kg of finished leather, while leaving over 600kg of solid wastes [3]. If not treated properly, these solid wastes will cause severe environmental pollution, and a huge waste of resources like high-quality collagen and chromic oxide [4]. As a result, a growing attention has been paid to the rational treatment and thorough utilization of these solid wastes, making the machining process clean and the products more valuable [5].

The existing studies on leather waste treatment mainly focus on extracting collagen and producing regenerated leather from the solid wastes [6]. One the one hand, collagen can be extracted either directly or indirectly from chrome shavings. The direct methods utilize chromium and collagen comprehensively, without separating them; the indirect methods separate and purify chromium and collagen, and makes use of them separately [7]. The indirect methods mainly include the following operations: pretreatment of chromiumcontaining wastes, dechroming and collagen extraction, separation and purification, concentration, drying and detection, and packaging. Dechroming is the key to collagen extraction. The dechroming effect directly bears on the product quality [8]. Based on the chromium remover or treatment agent, the collagen extraction strategies can be divided into acid method, alkali method, oxidation method, ultrasonic method, enzymatic method, acid-alkali alternating method, etc. [9]. Among them, the acid method deeply hydrolyzes collagen and only yields small molecules of collagen degradation products; what is worse, the equipment may be corroded by the acid, yet the chromium cannot be fully removed [10]. The alkali method features a long cycle, serious pollution, and large lime consumption; the collagen thus extracted is highly hydrolyzed, containing lots of ash and chromium [11]. The oxidation method produces toxic hexavalent chromium ions, which greatly harm the environment; the liquid waste of dechroming must be further treated [12]. The enzymatic method generates collagen hydrolysates with a small molecular weight, and requires dechroming in advance, failing to achieve a high removal rate of chromium; in addition, the collagen is severely degraded, the yield is low and the process is hard to control [13]. The collagens obtained by the above methods have mainly been applied in industries like leather, paper, food, medicine, cosmetics, feed, fertilizer, green fiber, etc. However, leather wastes contain so many harmful substances that they are not suitable for food, medicine, cosmetics or stock farming, even after precise purification and treatment.

On the other hand, the production of regenerated leather is inspired by paper-making. Firstly, various leather fibers are extracted from the shavings via plant tanning or chrome tanning, mixed with plant fibers like cotton, linen and wood fibers (the cotton and linen fibers mainly come from the cloth selvages and corners of sewing factories, the spinning wastes of textile factories, and various cotton and linen wastes [14]; the wood fibers are obtained from waste paper stock of paper mills and fiber pulp of low-grade paper), and then treated with chemical reagents and binders [15]. After that, a large amount of water is added, turning the mixture into a slurry. Next, the slurry is filtered by a metal sieve to remove water. Once the slurry is fully dried, it is possible to get laminated bonded fiberboards. The production process includes the following steps: selection of shavings and cotton fibers via screening and magnetic separation, weighing, pretreatment, crushing, rough grinding, fine grinding, slurry emulsification, preparation of mixed slurry, water removal by extrusion, forming, drying, sorting, and finishing. Apart from its high complexity, the production of regenerated leather generates wastewater containing various adhesives and compounding agents, causing secondary pollution to the environment [16]. Moreover, the finished products are not suitable for producing shoes, bags and similar items, because they are light in weight, weak in strength and elasticity, and poor in resistance to warping, temperature, wear and tear.

This paper attempts to recycle waste chrome shavings in a clean, simple and low-consumption manner. Firstly, the author selected the hot-pressing method to prepare genuine full-grain leather (GFGL) plates, and explored the influencing factors and their impacts of hot-press forming and the physicalmechanical properties of the preliminary plates, i.e. the hotpressing products of prefabricated plates. The exploration reveals that the main influencing factors include latex ratio (the ratio of latex weight to the total weight of shavings and latex), water ratio (the weight ratio of water to latex), hotpressing temperature, hot-pressing time and hot-pressing pressure. The range of each factor was optimized through single factor tests. Then, the optimal combination of these factors was determined through orthogonal test, coupled with the physical-mechanical properties and microstructure of the finished products.

2. METHODOLOGY

2.1 Main materials

(1) Chrome shavings. The wet blue chrome shavings obtained via cowhide production were selected for this research. The shavings were placed in a drying oven for 1h at 45° C to reduce the water content within 10%, making it possible to strictly control the subsequent screening and hot-pressing conditions

Considering their variation in size and shape, the directly recovered shavings were screened by different types of sieves. The screening shows that the shavings can be divided into the following categories: large strips of curled shavings, the shavings that can pass #8-mesh sieve, the shavings that can pass #30-mesh sieve, and that shavings that can pass #50-mesh sieve. Then, the author observed the microstructure of each type of shavings and tested the physical-mechanical properties of preliminary plates made of these shavings. It is learned that the curled shavings are too large, while the 30- and 50-mesh shavings are too fragmented. By contrast, the 8-mesh shavings have a medium size, and retain the original form of the collagenous fiber. In addition, the finished plates made of the 8-mesh shavings exhibited excellent physical-mechanical properties. Therefore, the 8-mesh shavings were taken as the raw materials.

Figures 1-4 present the macroscopic appearance and the microstructures magnified by 10,000, 5,000 and 2,000 times of the 8-mesh shavings, respectively. To ensure the consistency of the hot-pressed plates, every plate was prefabricated with 30g shavings.



Figure 1. The macroscopic appearance of the 8-mesh shavings

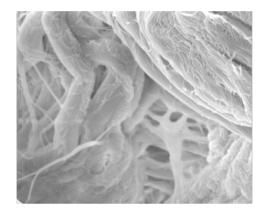


Figure 2. The microstructure of the 8-mesh shavings magnified by 10,000 times

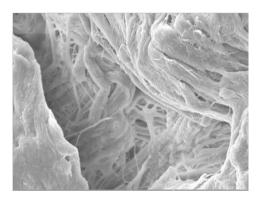


Figure 3. The microstructure of the 8-mesh shavings magnified by 5,000 times

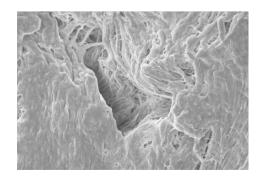


Figure 4. The microstructure of the 8-mesh shavings magnified by 2,000 times

(2) Natural rubber latex. This research uses the high ammonia (HA) latex, with a dry rubber content (DRC) of 60%, produced by Thai Hua Rubber Public Company Limited. The latex, fabricated in a clean and simple process, is a natural rubber adhesive with no toxicity, good compatibility with collagenous fiber, high hydrophilicity and stable performance. The main properties of the latex are listed in Table 1 below.

Table 1. The main properties of the latex

Properties	Results	Specifications as per ISO 2004: 1979E FG-HG		
		rg-l	16	
Total solids content (TSC) wt%	61.81	61.50	Min	
Dry rubber content (DRC) wt%	60.26	60.00	Min	
Non-rubber solids %	1.55	2.00	Max	
Ammonia content (on total weight) %	0.69	0.60	Min	
Ammonia content (on water phase) %	1.81	-	-	
pH at 25°C	10.45	-	-	
KOH number	0.62	1.00	Max	
Volatile fatty acid (VFA) number	0.028	0.20	Max	
Mechanical stability time@55% TS, sec.	866	650.00	Min	
Magnesium content on solids, ppm	21.11	80.00	ppm	

2.2 Hot-press forming process

(1) Prefabrication. The previous tests show that, after the latex was poured into the shavings, the two materials reacted quickly and bonded tightly in local areas. Hence, the plates were prefabricated in the following steps:

First, a proper amount of latex was poured into the beaker. Then, 30g 8-mesh shavings were added into the beaker. The mixture was stirred evenly with a glass rod immediately. After that, the mixture was relocated into a 33.5mm×33.5mm rigid frame mold, which was on a Teflon film. The mixture was flattened evenly with a glass rod, covered with a 33.5mm×33.5mm Teflon film, and then preformed through repeated punching with a hammer. The prefabricated plate was transferred onto the lower steel plate and covered with the upper steel plate, waiting to be hot-pressed.

(2) Hot-pressing. The prefabricated plate, sandwiched between the upper and lower steel plates, was placed on a flat vulcanizing machine, and hot-pressed at the test temperatures, pressures and times. The flat vulcanizing machine (model: XLB-D400*600*2/0.5MN) was produced by Qingdao Yahua Machinery Co., Ltd., Shandong Province, China.

2.3 Measurement of physical-mechanical properties

The focus of our test is the basic hot-pressing technique for GFGL plates made of natural rubber latex. Therefore, the tensile strength, bursting strength, gas permeability and water absorption of the preliminary plates were measured and characterized by the following instruments and methods.

The tensile strength, elongation at break, bursting strength and bursting height were tested on a universal testing machine (UTM). The gas permeability was captured by a H.C. Fedorov detector of leather air permeability. The mass percent of the absorbed water to the total specimen was measured with a kubelka apparatus after the plate had been soaked in the specified amount of water for 15min and 24h, respectively, at the required temperature.

2.4 Microstructure observation

The microstructure of the GFGL plates was observed by environmental scanning electron microscopy (ESEM) with the magnification of 3,000 and 6,000 times, respectively, to further reveal how the physical-mechanical properties of preliminary plates are affected by different levels of adhesive dosage, hot-pressing temperature, hot-pressing time and hotpressing pressure.

3. RESULTS ANALYSIS

3.1 Single factor test

3.1.1 The optimal ranges of hot-pressing factors based on formability, hot-pressing conditions, and appearance

(1) The optimal range of hot-pressing temperature. The results on samples $A1\sim3$ show that, after the addition of shavings, the HA natural rubber latex was clotted locally before stirring. To facilitate the processing and ensure the uniformity of preliminary plates, a certain proportion of purified water was added to the HA natural rubber latex.

The results on samples B1~3 demonstrate the significant impacts of hot-pressing temperature on the forming of the prefabricated plate, revealing that hot-press forming is infeasible at below 60°C. When the temperature was lower than 60°C, the natural rubber latex did not react with the shavings, which were dispersed and not bonded with the adhesive.

Then, the results on samples C1~4 were discussed to further explore the effects of high temperature on preliminary plates. It is learned that the prefabricated plates were hot-pressed into black circles, giving off the smell of burnt shavings, when the hot-pressing temperature reached 120°C.

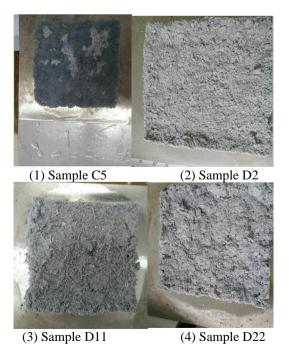


Figure 5. The photos on typical samples after hot-pressing

On this basis, the results at the hot-pressing temperature at 117 °C were examined, revealing that the prefabricated plates were also shaped like black circles, giving off the smell of burnt shavings. In addition, the results at 75 °C and 85 °C show that droplets remained on the press plate, while the preliminary plate stuck to the steel plates; the center of the preliminary plate was dry, the peripherals were wet blue, like the original shavings, and the shavings were scattered (Figure 5(1)). This is because the moisture cannot be completely volatized at such low hot-pressing temperatures.

Through the above analysis, the optimal range of hotpressing temperature was identified as $90{\sim}110$ °C.

(2) The optimal range of latex ratio. The results on D2, D5, D12, D17 and D22 show that the shavings were not well preformed at the latex ratio of 12%. Under this ratio, the

shavings were crumbs or floes and crumbs; the bluish color of the shavings was retained on the prefabricated plate, and some shavings fell off from the plate. When the latex ratio increased to $15\sim21$ %, the preformation improved, and the shavings existed as floes or floes and threads. When the latex ratio further grew to 23 %, the shavings combined widely into floes with a few threads or clots. In this case, the typical forms of the plates are shown in Figures 5(2)-(4).

Further analysis of C4 reveals that the plate was sticky and filaments appeared when the plate was bent. The filaments were rubber threads. Hence, the latex ratio of 25% is a little too high, and leads to a rise in cost.

In view of the above, the optimal range of the latex ratio was defined as $15\sim23$ %.

Table 2. The form	ability, hot-pres	sing conditions.	and appearance of	f samples

Sample number	Latex ratio, %	Water ratio	Hot-pressing temperature, °C	Hot-pressing time, min	Hot-pressing pressure, MPa	Formability of prefabricated plate	Hot-pressing conditions	Appearance of preliminary plate
						Rapid reaction and clots		Hard plate in local
A1	15	0; 1	90	3	3	in local areas, but loose	-	glued areas, and loose
						globally		shavings in unglued areas
						Rapid reaction, floes and		
A2	20	0:1	90	6	3	clots in local areas, but	Not hot-pressed	-
						loose globally Rapid reaction, floes and		
A3	25	0;1	90	6	3	clots in local areas, but	Not hot-pressed	-
		- 7				loose globally	· · · · ·	
B1	15	1:1	60	4	6	Loose but in shape	-	Not in shape
B2	20	1:1	90 Normal	2	6	Loose but in shape	-	In shape
B3	25	1:1	temperature	6	6	Loose but in shape	-	Dry and loose
C1	20	1:1	90	6	3	Floes	-	
C2	20	0.5: 1	120	6	3	Clots	-	Black, burnt smell
C3	25	1:1	120	6	3	Floes	-	Black, burnt smell,
C4	25	0.5: 1	90	6	3	Clots	_	sticky plate (Figure C3) Sticky plate
D1	12	0.5: 1	90	5	3	Crumbs, not in shape	-	-
D2	12	1:1	100	6	5	Floes and crumbs, in	_	_
D2	12	1.1	100	0	5	shape	-	-
D3	12	1.5: 1	110	7	7	Floes and threads, in shape	-	-
						Floes and threads, in		
D4	12	1.5: 1	100	6	3	shape	-	-
D5	15	0.5: 1	90	6	5	Floes, dry, in shape	-	-
D6 D7	15	1:1	100	7	7	Floes, in shape	-	-
D7 D8	15 15	1.5: 1 0.5: 1	110 100	5 7	3 3	Floes, in shape Floes, dry, in shape	-	-
D0 D9	15	1:1	110	5	5	Floes, in shape	_	-
D10	15	1.5: 1	90	6	7	Floes, in shape	-	Sticky steel plates
D11	15	1.5: 1	100	4	4	Floes, dispersed, but in	-	-
						shape Mixed floes and clots, in	A few droplets	In good shape,
D12	15	2:1	110	3	4	shape	on steel plates	compacted and dry
D12	15	2.5.1	100	5	4	Floes with a few	F	In good shape,
D13	15	2.5: 1	100	5	4	threads, in shape	-	compacted and dry
D14	18	1.5: 1	110	5	4	-	Droplets on steel plates	Compacted and hard plate
D15	18	2.5: 1	90	3	4	-	Droplets on press	Dry in the center and wet blue on the
				-	-		plate	peripherals
D16	21	0.5: 1	100	5	7	Floes, scattered, in shape	-	
D17	21	1:1	110	6	3	Floes with a few clots, in shape	-	-
D18	21	0.5: 1	110	6	7	Floes, scattered, in shape	-	-
D19	21	1:1	90	7	3	Floes with a few threads, in shape	-	-
D20	21	1.5: 1	90	5	5	Clots, in good shape	Lots of water on press plate Lots of water on	Sticky steel plates
D21	23	2:1	90	5	4	Dispersed, not in shape	press plate and steel plates	Good in shape
D22	23	2.5: 1	110	4	4	Clots with a few floes, best in shape	-	Slightly deformed, stiff and smooth

(3) The optimal range of water ratio. The results on samples $D5\sim13$ (latex ratio: 15 %) show that, when the water ratio was 0.5: 1, the shavings were dry and hard to be stirred evenly, while the plates were not well preformed. When the water ratio was 2.5:1, the shavings could be stirred evenly quickly without releasing heat, and the plates were well preformed. This means moisture promotes the bonding between shavings and latex. Of course, the temperature and time of hot-pressing should be increased to volatize the excess moisture.

The results on samples D16~22 (latex ratio: 18 % or 21 %) show that, when the water ratio was 0.5:1, the shavings were dispersed, and the plates were poorly preformed. When the water ratio was 2.5: 1, the shavings could be stirred evenly quickly, and the plates existed in the form of stable clots. In this case, the temperature and time of hot-pressing should also be increased to volatize the excess moisture.

In light of the above, the optimal range of water ratio was set to 1: $1 \sim 2.5$: 1.

(4) The optimal range of hot-pressing pressure. Comparing the results between B2 and A1, it is discovered that the hotpressing pressure affects the plate thickness and shape, but has nothing to do with the plate compactness. This is attributable to the good fluidity of natural rubber latex. Under a high pressure, the plates became thinner, changing from square shape to circular shape. This trend is particularly prominent in D3 and D6: the plates were thinned and severely deformed at the pressure of 7MPa and at the time of 7min. Hence, 7MPa is too large for hot-press forming. Therefore, the optimal range of hot-pressing pressure was determined as 3~6MPa.

(5) The optimal range of hot-pressing time. Results analysis shows that the hot-pressing time had insignificant impacts on the physical-mechanical properties of the preliminary plates. However, it takes time for the moisture to volatize in the hot-pressing process. Hence, the optimal range of hot-pressing time was set to $4\sim7$ min.

3.1.2 The optimal range of hot-pressing factors based on the physical-mechanical properties

The physical-mechanical properties of the preliminary plates with relatively good appearance in Table 2 were measured. The measured results on tensile strength, elongation at break, gas permeability, 15min water absorption and 24h water absorption is displayed in Figures 6~10, respectively.

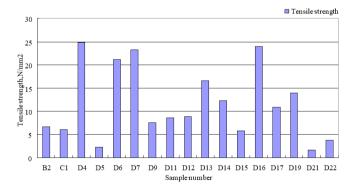


Figure 6. Tensile strength

As shown in Figure 6, D4, D6, D7, D13, D16 and D19 had relatively high tensile strength, while D5, D15, D21 and D22 had relatively poor resistance to tensile force. Thus, it can be concluded that the tensile strength is affected by the interaction between the latex ratio, water ratio, hot-pressing temperature, hot-pressing time and hot-pressing pressure; the minimum tensile strength appears at the latex ratio of 23 %.

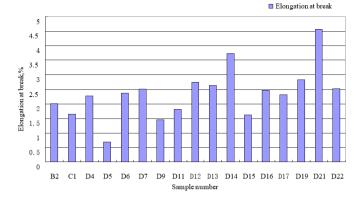


Figure 7. Elongation at break

As shown in Figure 7, D21 and D14 had relatively high elongations at break, while D5 came at the bottom in terms of this factor. The result reflects the impacts of water ratio on elongation at break: the higher the water ratio, the greater the elongation at break.

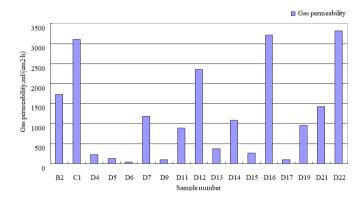


Figure 8. Gas permeability

As shown in Figure 8, good gas permeability was observed in C1, D12, D16 and D22, and poor gas permeability was found in D4, D5, D6, D9, D13, D15 and D17. Therefore, the gas permeability of the samples is affected by the interaction between the latex ratio, water ratio, hot-pressing temperature, hot-pressing time and hot-pressing pressure.

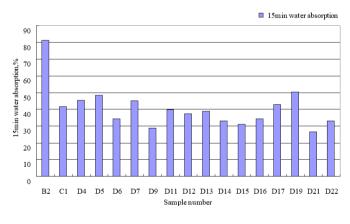


Figure 9. 15min water absorption

As shown in Figures 9 and 10, B2 was the only sample with a good water absorption at 15min, while the other samples were similar on water absorption. This proves that the rapid water absorption of the samples is not influenced by hotpressing factors. However, with the elapse of time, the water absorption of B2, D11 and D17 declined, while D5 had the same water absorption at 24h as at 15min. The results demonstrate that the prefabricated plates can absorb water rapidly.

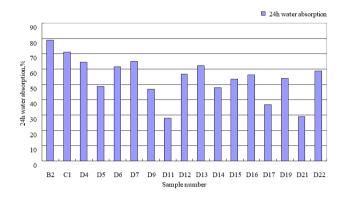


Figure 10. 24h water absorption

The above analysis on physical-mechanical properties shows that the optimal range of hot-pressing factors, which were determined based on the formability, hot-pressing conditions and appearance agree well with the properties of the preliminary plates.

3.2 Orthogonal test

An orthogonal test with 5 factors and 4 levels was designed based on the optimal ranges of hot-pressing factors (Table 3).

3.2.1 Physical-mechanical properties analysis

The physical-mechanical properties of the preliminary plates with relatively good appearance in Table 3 were measured. The measured results on tensile strength, elongation at break, bursting strength, bursting height, gas permeability, 15min water absorption and 24h water absorption is displayed in Figures 11~17, respectively.

Table 3.	The	5-factor,	4-level	orthogonal	test
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Sample number	Latex ratio, %	Water ratio	Hot-pressing temperature, °C	Hot-pressing time, min	Hot-pressing pressure, MPa
S-1	15	1:1	90	4	3
S-5	15	2:1	110	6	5
S-10	15	2.5:1	115	7	6
S-9	18	1:1	100	6	6
S-11	18	1.5:1	90	7	5
S-12	18	2:1	115	4	4
S-13	18	2.5:1	110	5	3
S-14	21	1.5:1	115	6	3
S-2	21	2:1	90	5	6
S-6	21	2.5:1	100	4	5
S-4	23	1.5:1	110	4	6
S-7	23	1:1	115	5	5
S-3	23	2:1	100	7	3
S-8	23	2.5:1	90	6	4

25

20

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Bursting 10

0

S-1

S-2

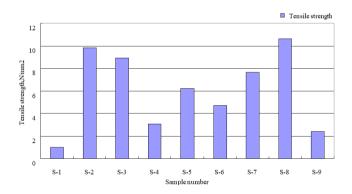


Figure 11. Tensile strength

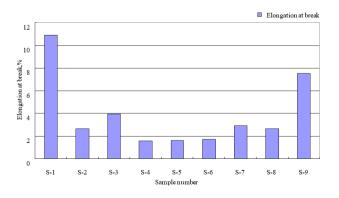


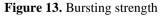
Figure 12. Elongation at break

S-3 S-4 S-5 S-6 S-7 Sample number

Bursting strength

S-8

S-9



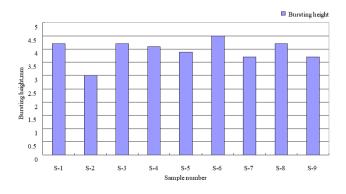


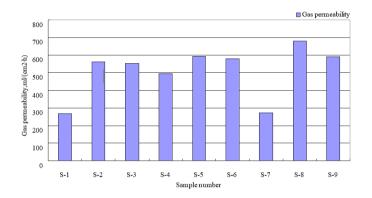
Figure 14. Bursting height

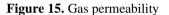
As shown in Figure 11, S-1 and S-9 had relatively low tensile strengths, which agrees with the phenomenon that the preliminary plate stuck to the steel plate and some shavings fell of the plate. Through the analysis on hot-pressing factors, the phenomenon can be explained by the fact that the water ratio of 1:1 is too low for the latex to be mixed evenly with the shavings.

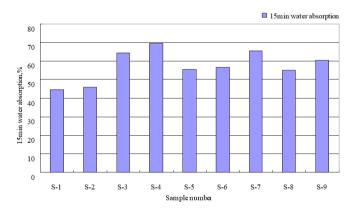
It can also be seen that the GFGL plates made of natural rubber latex showed relatively low tensile strength. Hence, tough plant fibers will be added to enhance the tensile strength.

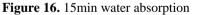
Figure 12 shows that the elongation at break of the GFGL plates made of natural rubber latex is negatively correlated with their tensile strength.

Figures 13 and 14 show that the GFGL plates made of natural rubber latex boast good bursting strength, which is basically not affected by hot-pressing factors. It can be concluded that such plates are excellent materials for insole board.









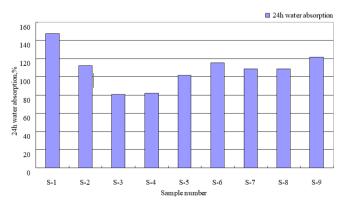


Figure 17. 24h water absorption

As shown in Figure 15, all samples exhibited good gas permeability except S-1 and S-7. The sample S-1 had poor gas permeability, because the plate was not well formed due to low latex ratio, water ratio and hot-pressing temperature. Despite its high latex ratio, the sample S-7 still had a low gas permeability, for the plate was not well formed under the low water ratio, which hinders the uniform mixing of the latex and the shavings.

Figures 16 and 17 show that the GFGL plates made of natural rubber latex enjoy excellent water absorption, which improves with the elapse of time. This is because natural rubber latex and shavings are both hydrophilic materials. After hot-pressing, the network structure of the two materials is bound to have a high-water absorption. Thus, the GFGL plates made of natural rubber latex are desirable materials for insole board.

3.2.2 Microstructure observation

Figures 18~21 present the SEM images of typical samples S-1, S-2, S-3 and S-9, respectively, with the magnification of 3,000 times.

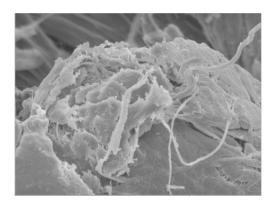


Figure 18. The SEM image of S-1

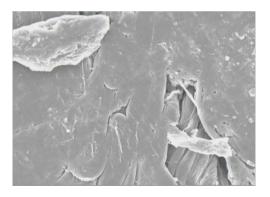


Figure 19. The SEM image of S-2

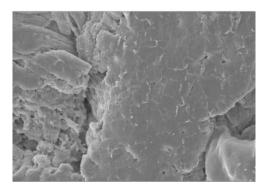


Figure 20. The SEM image of S-3

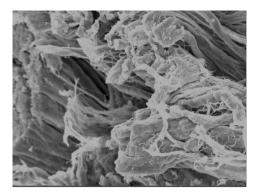


Figure 21. The SEM image of S-9

In the above figures, S-2 and S-3 showed integrated network structures, which was formed through the intense chemical reactions between the natural rubber latex and the shavings under stirring and hot-pressing. By contrast, the latex and the shavings were not well bonded in S-1 and S-9. The results are consistent with the analysis on tensile strength and gas permeability.

Considering the SEM results and the orthogonal test, the hot-pressing parameters of S-5 and S-6 were determined as the optimal settings for basic hot-pressing.

4. CONCLUSIONS

The above analysis shows that it is feasible to produce GFGL plates through hot-pressing on a flat vulcanizing machine. This dry machining method is simple and cost-effective, and can effectively recycle chrome shavings, without causing any secondary pollution to the environment.

The optimal hot-pressing conditions and formula for GFGL plates made of natural rubber latex were determined as follows: (1) latex ratio 15 %, water ratio 2:1, 110 °C, 6min and 5MPa; (2) latex ratio 21 %, water ratio 2.5:1, 100 °C, 4min and 5MPa.

Under these settings, the preliminary plates enjoy good gas permeability and water absorption, and enough tensile strength and rupture strength. If toughed by plant fibers, these plates can be adopted to produce insole board in footwear industry.

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